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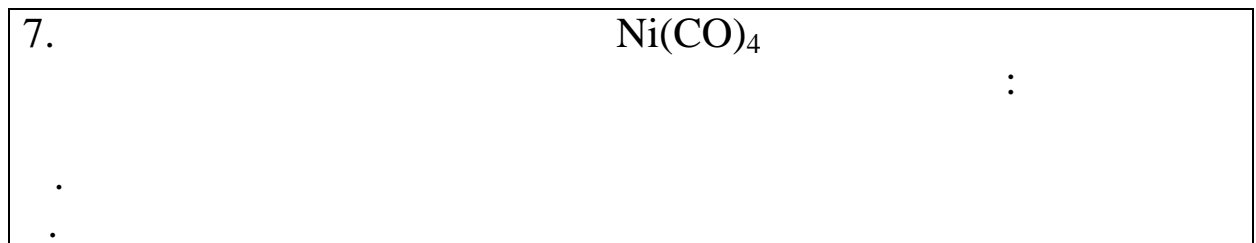
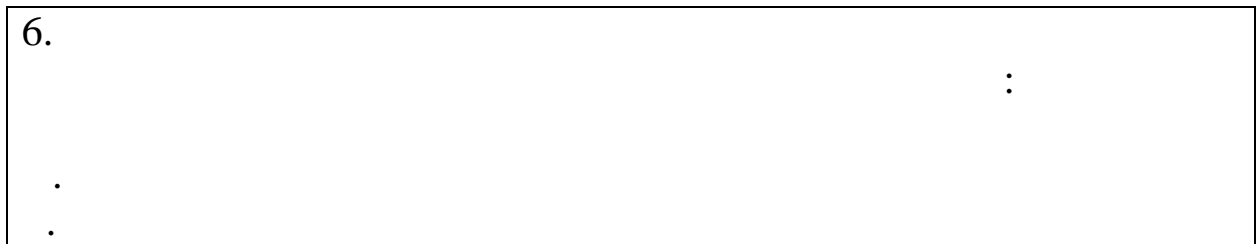
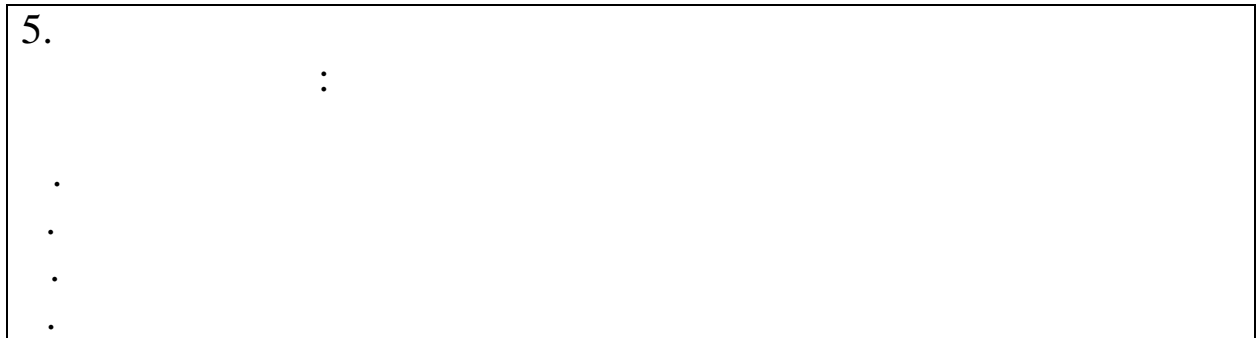
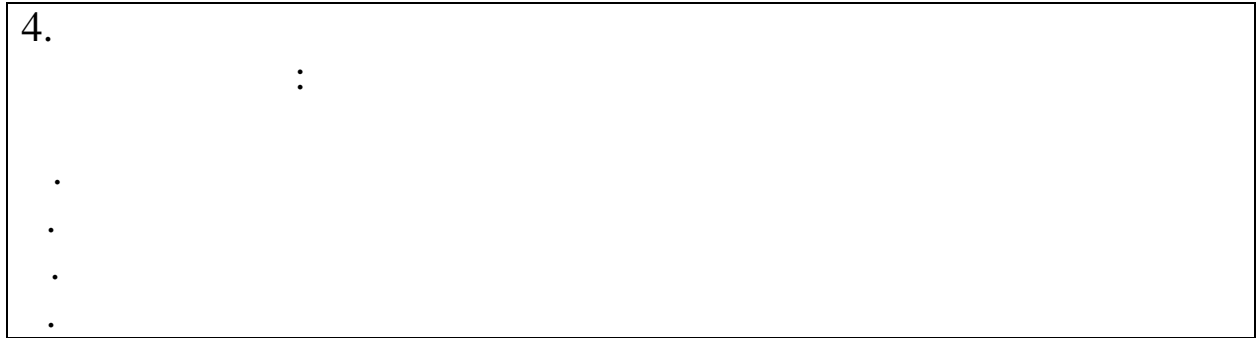
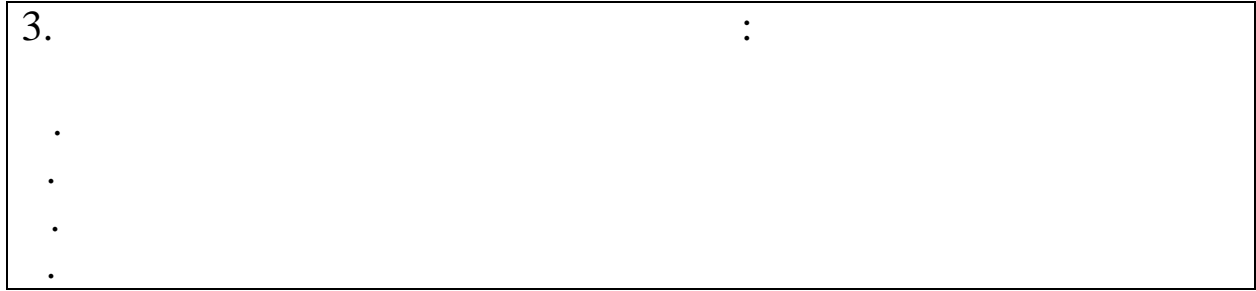
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12.

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- . CH₃OH
- . C₆H₆
- . (C₂H₅)₂O
- . (CH₃)₂O
- . C₆H₅NO₂
- . C₆H₅NH₂

13.

:

- . C₂H₅OH
- . C₆H₆
- . (CH₃)₂O
- . (CH₃)₂CO
- . C₆H₅CH₃
- . C₆H₅NO₂

14.

:

- . N_2O
- . $\text{C}_2\text{H}_5\text{Cl}$
- . C_6H_6
- . $\text{C}_2\text{H}_5\text{OH}$
- . $\text{C}_6\text{H}_5\text{NH}_2$
- . $\text{C}_6\text{H}_5\text{CH}_3$
- . $\text{C}_2\text{H}_5\text{-O-C}_2\text{H}_5$

15.

:

- . $\text{C}_2\text{H}_5\text{-O-C}_2\text{H}_5$
- . C_6H_6
- . $\text{C}_6\text{H}_5\text{OH}$
- . $\text{C}_3\text{H}_7\text{Br}$
- . $\text{OH-C}_6\text{H}_4\text{-OH}$
- . CH_3OH
- . $\text{CH}_3\text{-CO-CH}_3$

16.

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|---|-----------------|---|
| . | | . |
| . | Z_{ch} | . |
| . | Z_{ac} | . |
| . | | . |

17.

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18.

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-
- Z_{ac}
- Z_{ch}
- Lim_{ac}
- Lim_{ch}
- C_{20°
- 50

19.

:

,

- 50
- /
- C_{20°
-
-
- Z_{ac}
-
- 50
- Lim_{ac}
- Z_{ch}

20.

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- 50
- $\frac{1}{50}$
- 50
- $\frac{1}{50}$

21.

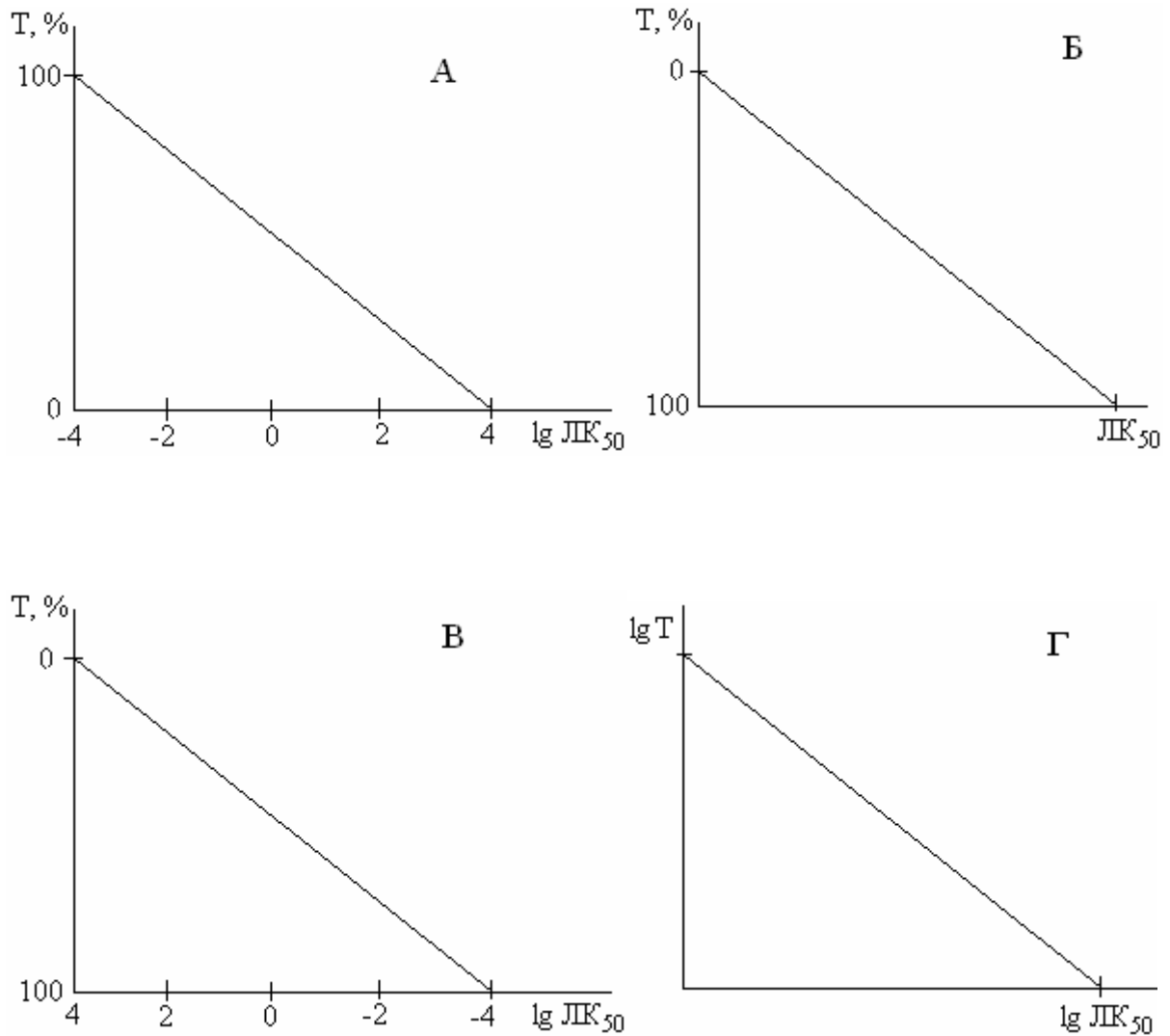
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- ..
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-
- Lim_{ac}
- Lim_{ch}
- C_{20° ;
-
- /
- 50
- Z_{ac}

22.

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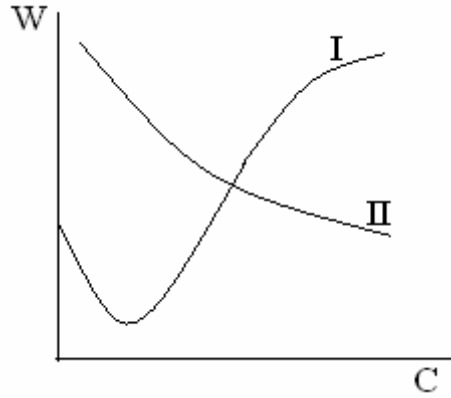


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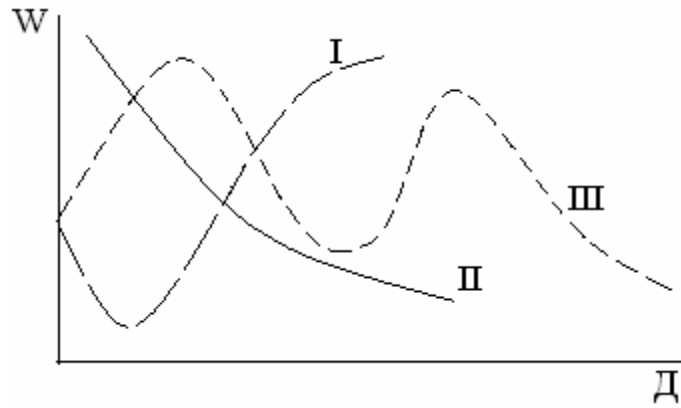
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$$\begin{aligned} & \cdot \frac{Z_{ch} \cdot}{Z_{ac}} & \cdot \frac{a \cdot Z_{ch} \cdot}{Z_{ac}} \\ & \cdot \frac{a \cdot Z_{ac} \cdot}{Z_{ch}} & \cdot \frac{a \cdot Z_{ch}}{Z_{ac} \cdot} \\ & \cdot \frac{a \cdot Z_{ac}}{Z_{ch}} \end{aligned}$$

24.
(W –):
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25.
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. 2, 3, 5, 7, 8

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. H₂

. HCN

. N₂

4.

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_____ $\frac{\text{SiO}_2}{\%}$ _____

- . 1. < 2 I.
- . 2. > 2 II.
- . 3. = 2 III.
- . 4. = 2 IV.
- V.

5. _____ :

SiO₂, %

.	1. = 2	I. _____
.	2. > 2	II. _____
	3. = 2	III. _____
	4. > 2	IV. _____
	5. < 2	V. _____
		VI. _____

6. _____ :

.	_____	1. _____
.	_____	2. _____
		3. _____
		4. _____ / _____
		_____ / _____

7. _____ :

_____	_____ - _____	_____ / _____
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8.

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.I.
II
III.

9.

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_____:

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. Z_{ac} . Z_{ch}

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I. $\frac{C}{C_0}$

II. $\frac{C_{20^0}}{50}$

III. $\frac{Lim_{ch}}{}$

IV. $\frac{C_{20^0}}{50}$

V. $\frac{Lim_{ac}}{Lim_{ch}}$

VI. $\frac{50}{Lim_{ac}}$

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10. , :

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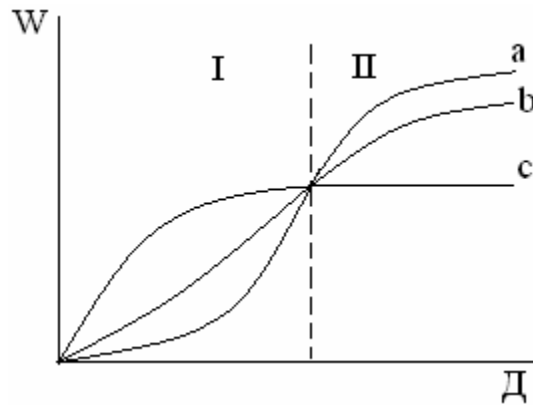
I.

II.

(: , ,)

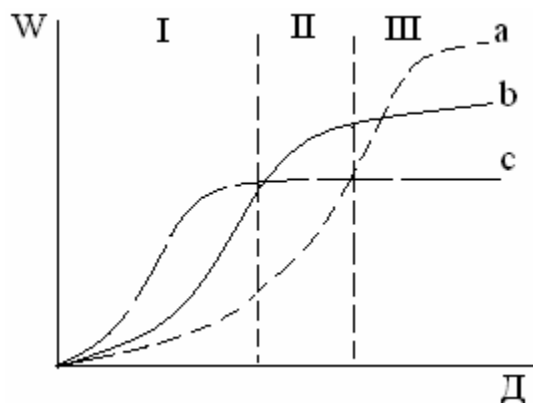
1.

I II (W -):



2.

I, II, III (W -):



1.6

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«) »

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1.

(TOX.)

(PHYS.):

- . $\text{Lim}_{\text{ch}} > \text{Lim}_{\text{ac}}$
- . $\text{Lim}_{\text{ac}} > \text{Lim}_{\text{ch}}$
- . $\text{Lim}_{\text{phys}} > \text{Lim}_{\text{tox}}$
- . $\text{Lim}_{\text{tox}} > \text{Lim}_{\text{phys}}$

2.

:

- . $\text{Lim} < \text{Lim}$
- . $\text{Lim} > \text{Lim}$

3.

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· ,

4.

0,3 / ³
(). ,
3

· $C_{III} = \frac{C_{II}}{3} = 0,1 \quad / \quad ^3$

· $C_{III} = C_{II} \cdot 3 = 0,9 \quad / \quad ^3$

5.

0,45 / ³
(v).
1,5.

· $C_{III} = \frac{C_{IV}}{1,5} = 0,3 \quad / \quad ^3$

· $C_{III} = C_{IV} \cdot 1,5 = 0,675 \quad / \quad ^3$

6.

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1.

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. 1-100	II.
. 100-10000	III.
. > 10000	IV.

2.

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. C_{20°

. /

I. $\frac{p \cdot M}{18,3}$

II. $\frac{62,3 \cdot S \cdot T}{p \cdot M}$

III. $\frac{C_{20^\circ}}{50}$

IV. $\frac{C_{20^\circ}}{50}$

V. $\frac{p^2 \cdot M^2}{S \cdot 50}$

3.

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1.

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4.

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I.

II.

III. ,

IV.

V.

VI.

VII. -

VIII. SH-

IX.

X.

XI.

XII.

XIII.

5.

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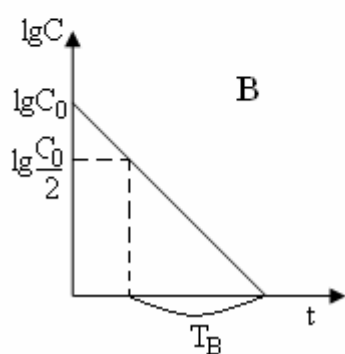
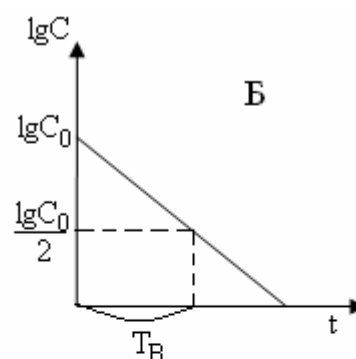
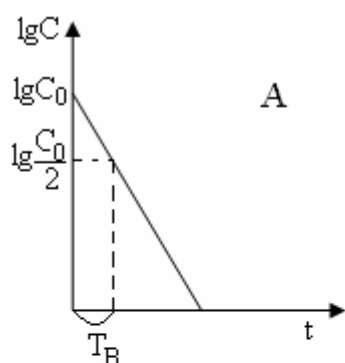
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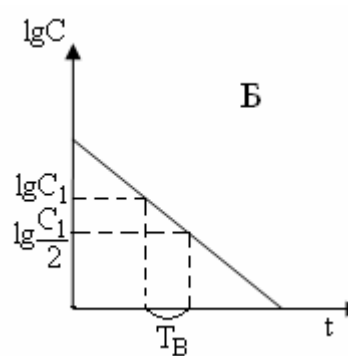
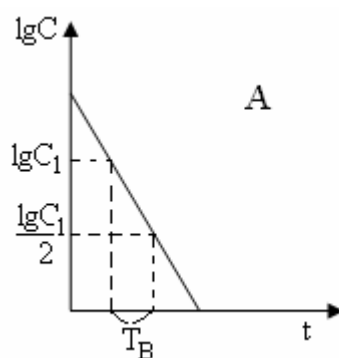
1. _____

2. _____

3.



4.



5.

- ,
- . $V < 14$
 - . $14 < V < 42$
 - . $V > 42$
 - . $V = 10$

6.

- ,
- . $V < 10$
 - . $V = 10$
 - . $V < 14$
 - . $14 < V < 42$
 - . $V > 42$

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(

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1.

1. α

$$\frac{0,693}{T}$$

2. V

$$\frac{\ln 2}{T}$$

3. k

$$\frac{D}{C}$$

2.

1. V

$$\cdot \frac{\ln 2}{T}$$

2. k

$$\cdot \frac{D - W_t}{C}$$

3. α

$$\cdot \frac{\ln 2}{T}$$

3.

1. α

$$\cdot \frac{C - V}{C}$$

2. k

$$\cdot \ln 2 \frac{V}{K}$$

3. T

$$\cdot \frac{D}{C_0}$$

4. T

$$\cdot \frac{\ln 2}{T}$$

5. V

$$\cdot \frac{\ln 2}{T}$$

6. K

$$\cdot \frac{\ln 2}{x}$$

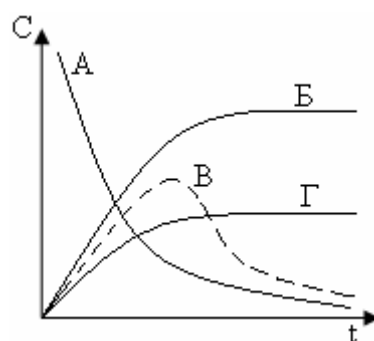
4.

$$1. C = C_0 \cdot e^{-xt}$$

$$2. C = \frac{C_0 \lambda}{2} (1 - e^{-2kt})$$

$$3. Y = \frac{Dk}{x-k} (e^{-kt} - e^{-xt})$$

$$4. C = C_0 \lambda (1 - e^{-kt})$$



5.

1.

$$C = \frac{C_0 \lambda}{2} (1 - e^{-2kt})$$

I. $C_0 \lambda$

2.

$$C = \frac{C_0 \lambda k}{x} (1 - e^{-xt})$$

II. $\frac{C_0 \lambda}{2}$

$$C = C_0 \lambda (1 - e^{-kt})$$

III. $\frac{k}{x} \lambda C_0$

6.

1.

$$\frac{C_0 \lambda}{2}$$

I.

2.

$$\frac{k}{x} \lambda C_0$$

II.

$$C_0 \lambda$$

7.

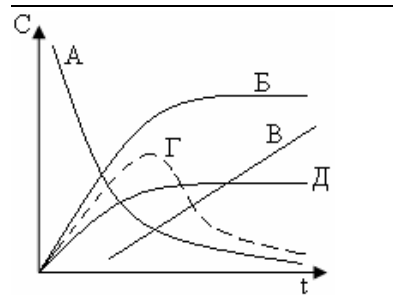
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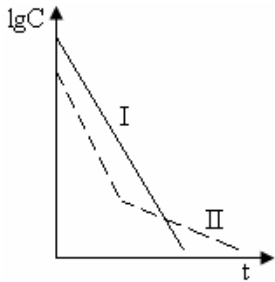
5.



8.

1. $C = C_{0_1} \cdot e^{-x_1 t} + C_{0_2} \cdot e^{-x_2 t}$

2. $C = C_0 \cdot e^{-(x_1 + x_2) t}$



The graph shows a vertical axis labeled 'lg C' and a horizontal axis labeled 't'. Two curves, labeled I and II, originate from the same point on the vertical axis. Curve I is a straight line with a negative slope. Curve II is a curve that starts with a steeper negative slope than curve I but eventually levels off, becoming less steep than curve I as time increases.

9.

1. Y $\cdot D \left(\frac{k}{x} \right)^{\frac{x}{x-k}}$

2. t_{\max} $\cdot \frac{Dk}{x-k} (e^{-kt} - e^{-xt})$

3. Y_{\max} $\cdot \frac{1}{k-x} \ln \frac{k}{x}$

:

(« »)

1. CS_2

$$C = 8,66 \cdot e^{-0,46t} + 0,052 \cdot e^{-0,035t}$$

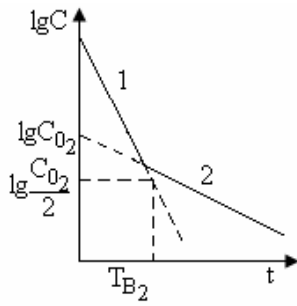
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1. ,

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$\lg C - t?$



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1.

1.

2.

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2.

1.

$$. C = C_0 \cdot e^{-xt}$$

2.

$$. C = C_0 - \frac{xt}{2,3}$$

$$. I = I_0 - \frac{xtm}{2,3}$$

$$. I = I_0 \cdot e^{-xt}$$

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(« »)

1.

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2.

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$$. C = C_0 \cdot e^{-(x_1+x_2)t}$$

$$. C = C_0 \cdot e^{-xt}$$

$$. C = C_{0_1} \cdot e^{-x_1t} + C_{0_2} \cdot e^{-x_2t}$$

3.

?

$$. C = C_0 \cdot e^{-(x_1+x_2)t}$$

$$. C = C_0 \cdot e^{-xt}$$

$$. C = C_{0_1} \cdot e^{-x_1t} + C_{0_2} \cdot e^{-x_2t}$$

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2.2

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1.			
	_____	_____	_____
1.	.		I.
2.	.		II.
			III.
			IV.
			V.
			VI.
			VII.

2.

1. _____ ,
 2. _____

I.
 II.

3.

1. _____
 2. _____

I.
 II.

4.

1. _____
 2. I
 3. $1/$

$\cdot \frac{50}{\sum 50}$
 $\cdot \frac{50}{\sum 50}$
 $\cdot 1 - \frac{2}{1}$

I.
 II.

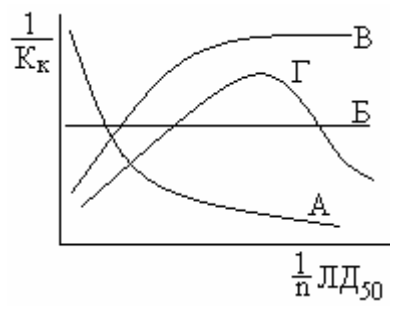
5.

1. _____
 2. _____
 3. _____

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6. ()

1. 2. 3. 4.



The graph shows the relationship between the inverse of the coefficient of utilization, $\frac{1}{K_K}$, and the inverse of the average utilization factor, $\frac{1}{n} \Pi Д_{50}$. Three curves are plotted: Curve A starts high and decreases; Curve B starts low and increases to a plateau; Curve Γ starts low, increases to a peak, and then decreases. A horizontal line is drawn across the graph.

7.

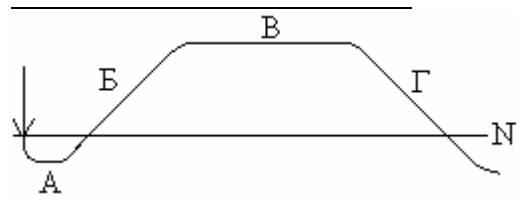
1. ;

2. ,

3. .

8.

1. 2. 3. 4.



The graph shows a trapezoidal curve with points A, B, Γ, and N. Point A is at the start of a downward slope, B is at the top of a horizontal plateau, Γ is at the end of a downward slope, and N is at the end of a horizontal line. A vertical arrow points down at point A.

(:)

1.

1.

2.

3.

4.

5.

6.

7.

8.

:

- 1, 3, 5, 7

- 2, 4, 6, 7

- 2, 4, 6, 8

- 1, 4, 6, 7

- 1, 4, 6, 8

2.

1.

2.

3.

4.

5.

6.

7.

8.

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- 2, 3, 5, 8

- 2, 4, 6, 8

- 1, 4, 6, 7

- 1, 4, 5, 7

3.

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4.

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5.

(II)

1.

2.

3.

6.

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I.

II.

7.

50 %

25

15-

1/20 50

$$\cdot \frac{\frac{1}{20} \cdot 50}{50} = \frac{1,25}{25} = 0,05$$

$$\cdot \frac{50}{\frac{1}{20} \cdot 50} = \frac{25}{1,25} = 20$$

$$\cdot \frac{15 \cdot \frac{1}{20} \cdot 50}{50} = \frac{15 \cdot 1,25}{25} = 0,75$$

8.

50 %

12 /

10-

1/5 50

$$\cdot \frac{\frac{1}{5} \cdot 100\%}{50} = \frac{12 \cdot 100\%}{2,4} = 500\%$$

$$\cdot \frac{10 \cdot \frac{1}{5} \cdot 50}{50} = \frac{10 \cdot 2,4 \cdot 100\%}{12} = 200\%$$

$$\cdot \frac{50 \cdot 100\%}{10 \cdot \frac{1}{5} \cdot 50} = \frac{12 \cdot 100\%}{10 \cdot 2,4} = 50\%$$

9.

50 %

20 /

15

1,6 /

$$\cdot 1 + \frac{1}{2} = 1 + \frac{20}{1,6} = 13,5; \quad \cdot 1 - \frac{2}{1} = 1 - \frac{1,6}{20} = 0,92; \quad \cdot 1 + \frac{2}{1} = 1 + \frac{1,6}{20} = 1,08.$$

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1.

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2.6

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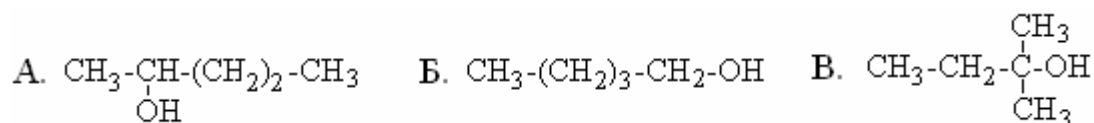
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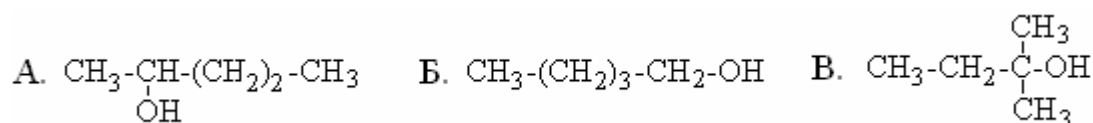
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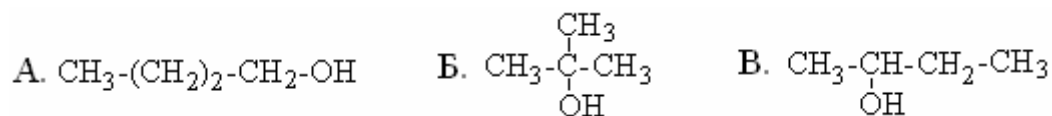
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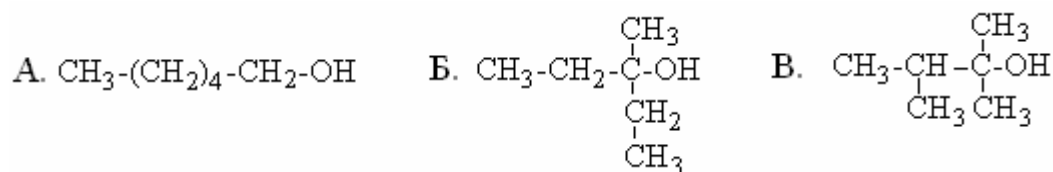
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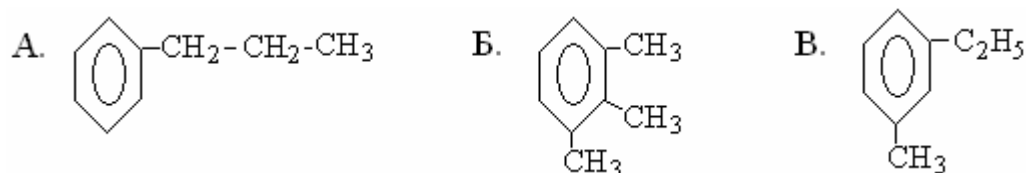
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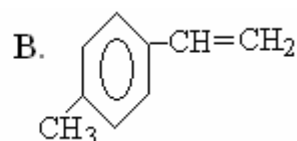
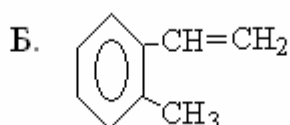
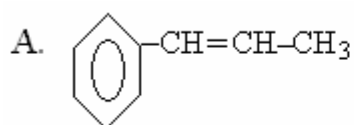
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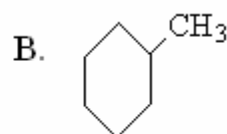
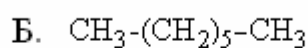
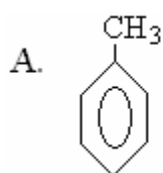
11.



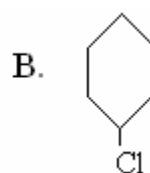
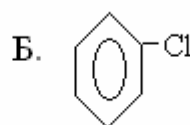
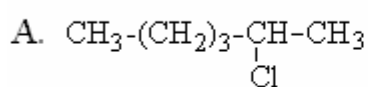
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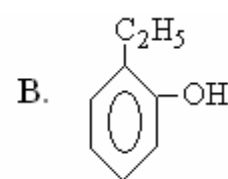
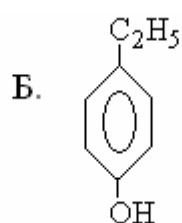
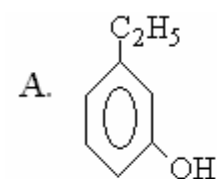
13.



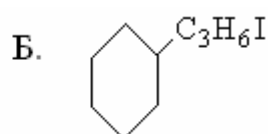
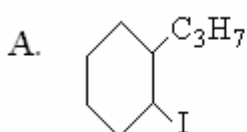
14.



15.



16.



20.

- . $C_6H_4Cl_2$
- . C_6H_5Cl
- . $C_6H_3Cl_3$

I.

II.

:

(, , ,)

1. ,

1.

2.

. C_3H_6ClOH . C_3H_7OH . CH_3-O-CH_3 . $CH_2Br-O-CH_3$

2. ,

1.

2.

. $CH_2=CH-CH=CH_2$. $CH_2=CCl-CH=CH_2$. $C_2H_5-O-C_2H_4Cl$. $C_2H_5-O-C_2H_5$

3. ' ,

1. _____

. $\text{CCl}_2=\text{CHCl}$

2. _____

. $\text{CH}_2=\text{CH}_2$. $\text{C}_4\text{H}_9\text{OH}$. $\text{C}_4\text{H}_8\text{BrOH}$

4.

. $\text{C}_6\text{H}_5\text{NH}_2$. $\text{C}_6\text{H}_5\text{NHCOCH}_3$

,

I. -

II.

III. -

5.

. $\text{C}_6\text{H}_5\text{NHCH}_2\text{COOH}$. $\text{C}_6\text{H}_5\text{NHCH}_3$

I. -

II. -

III.

6.

- . -OH
- . -CO-CH₃
- . -NO₂
- . -NO
- . -COO
- . -NH₂

7.

- 1.
- 2.

- . C₃H₇-O-NO
- . C₆H₅-NO₂
- . CH₂ONO₂-CHONO₂-CH₂ONO₂
- . C₂H₅-O-NO₂
- . C₆H₄(NO₂)₂
- . C₆H₅NH₂

8.

- 1.
- 2.

- . C₆H₄(NO₂)NH₂
- . C₃H₇-O-NO
- . CH₂ONO₂-CHONO₂-CH₂ONO₂
- . C₆H₄(NO₂)₂

9. ' _____

1. _____

2. _____

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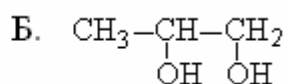
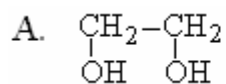
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1.



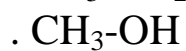
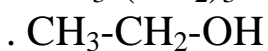
2.



3.

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4.



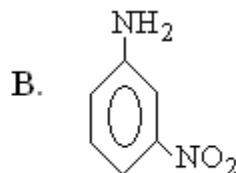
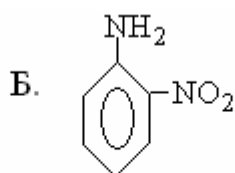
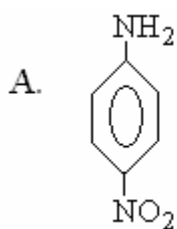
5.

1.

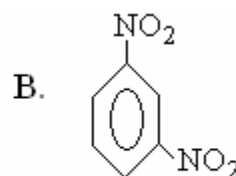
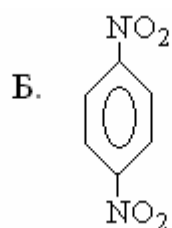
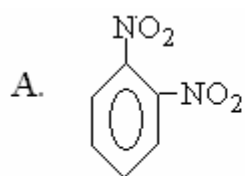
2.



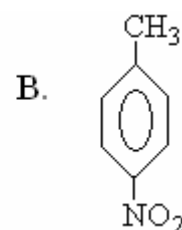
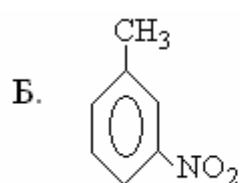
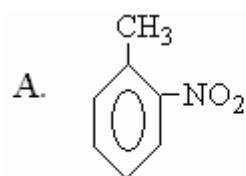
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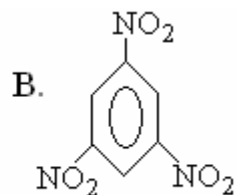
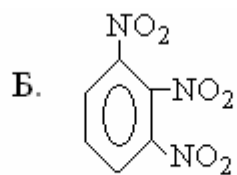
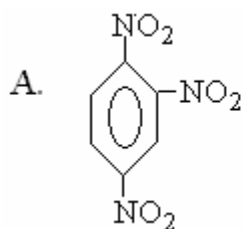
7.



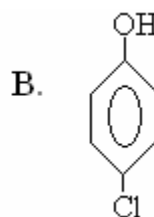
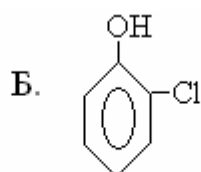
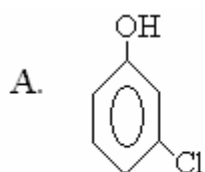
8.



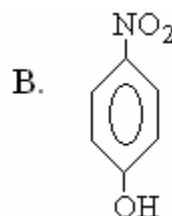
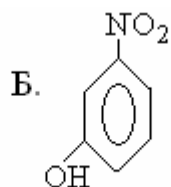
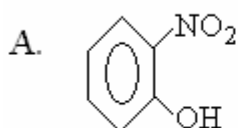
9.



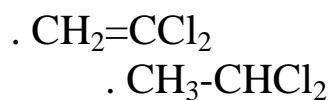
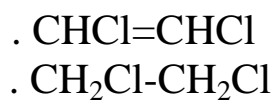
10.



11.



12.



13.

1.

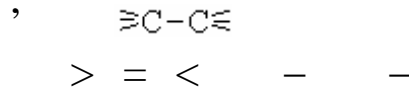
2.

3.

4.

5.

6.



:

- 1, 3, 5

- 2, 4, 6

- 1, 4, 6

- 1, 4, 5

- 2, 3, 5

14.

1.

2.

3.

4.

5.

:

- 1, 2, 4

- 1, 2, 5

- 3, 5

- 3, 4

- 2, 4

- 2, 5

15.

1. -COOH
2. -NH₂
3. -NO
4. -NO₂
5. -COOH-CH₃

:

- 1, 2, 3

- 2, 3, 5

- 2, 3, 4

- 1, 3, 4

- 1, 3, 5

16.

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17.

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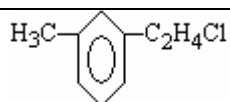
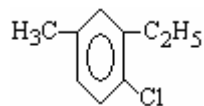
(_____ - _____)

1.

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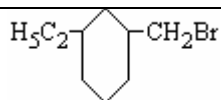
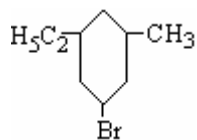
_____ 1 _____ 2 _____ ,

2.

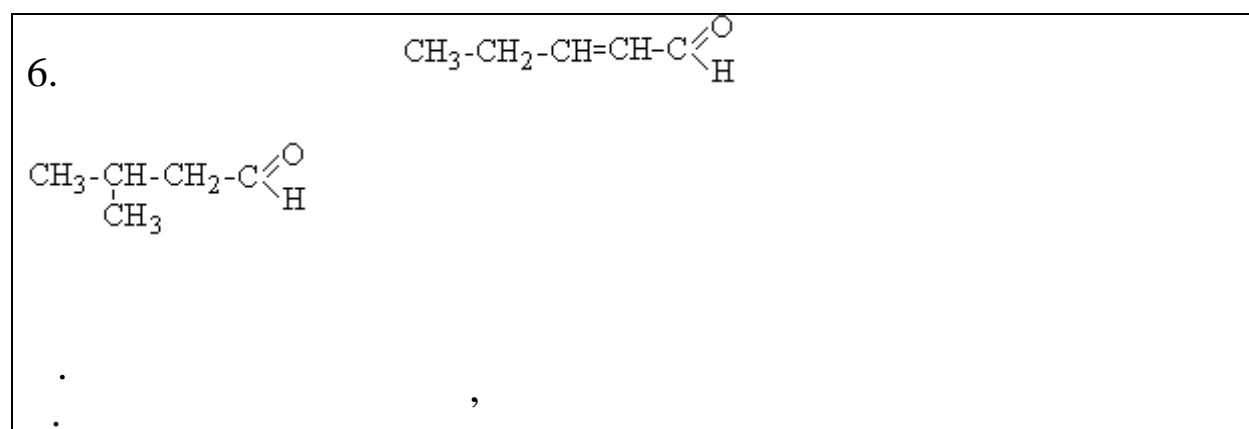
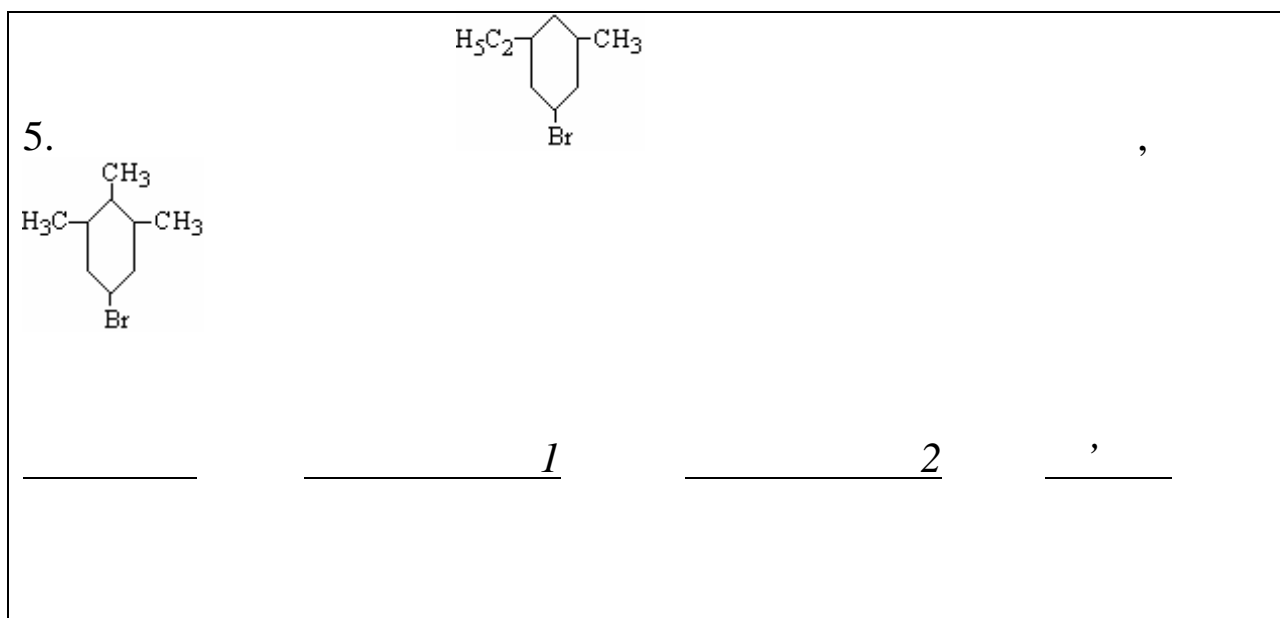
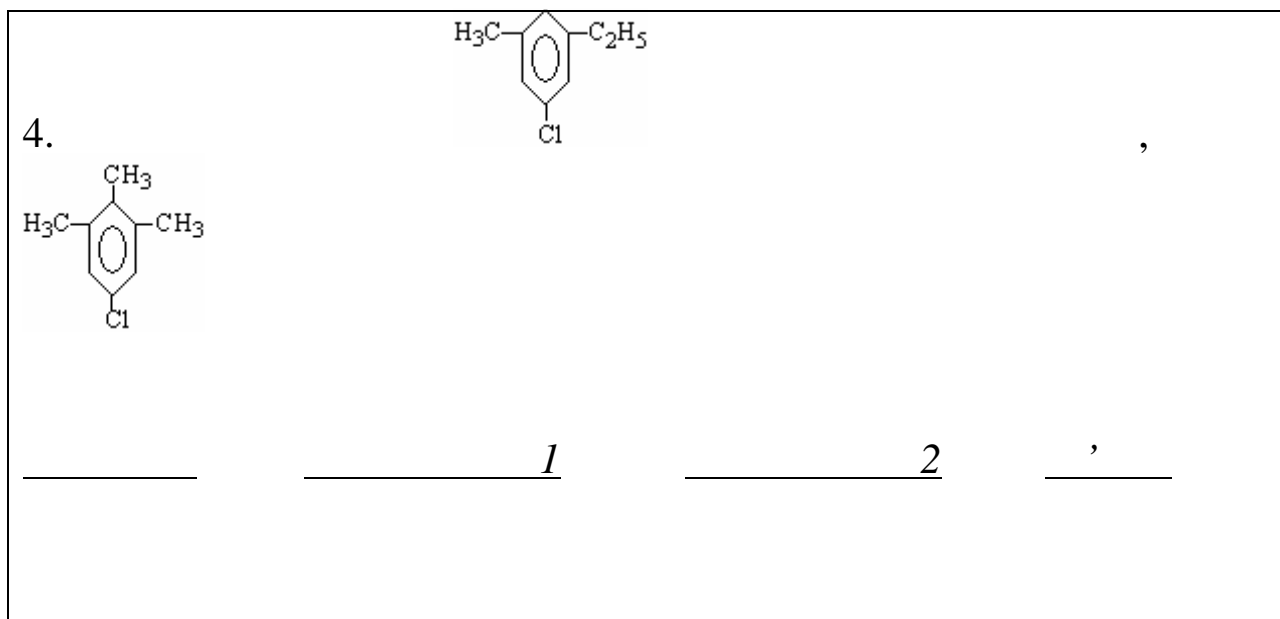


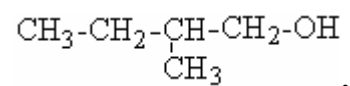
_____ 1 _____ 2 _____ ,

3.



_____ 1 _____ 2 _____ ,

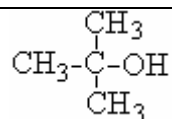




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8.

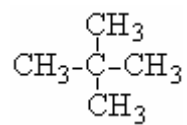


- . $\text{CH}_3-(\text{CH}_2)_2-\text{CH}_2\text{OH}$
- . $\text{CH}_3-\text{CH}_2-\text{CHCl}-\text{CH}_2\text{OH}$

1.

2.

9.



- . $\text{CH}_3-(\text{CH}_2)_3-\text{CH}_3$
- . $\text{CH}_3-\text{CH}=\text{CH}-\text{CH}_2-\text{CH}_3$

1.

2.

(: , ,) , ,

1.

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2. _____ .

3. _____ .

4. _____ .

5. _____ .

(: « »)

1.

· CH₃COOH
· HCOOH

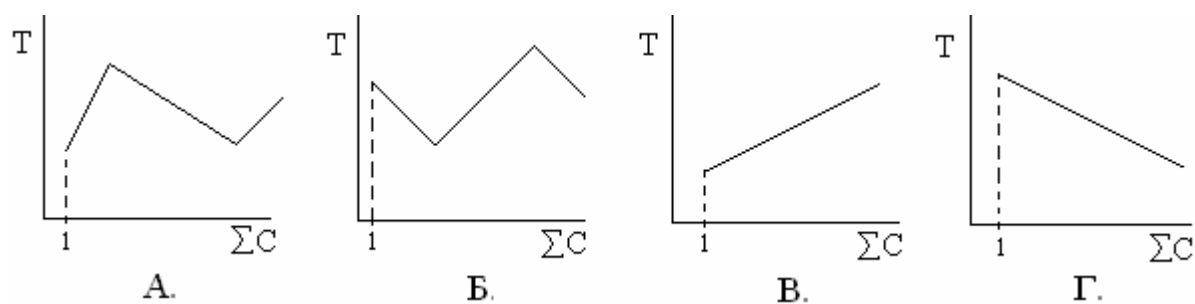
2.

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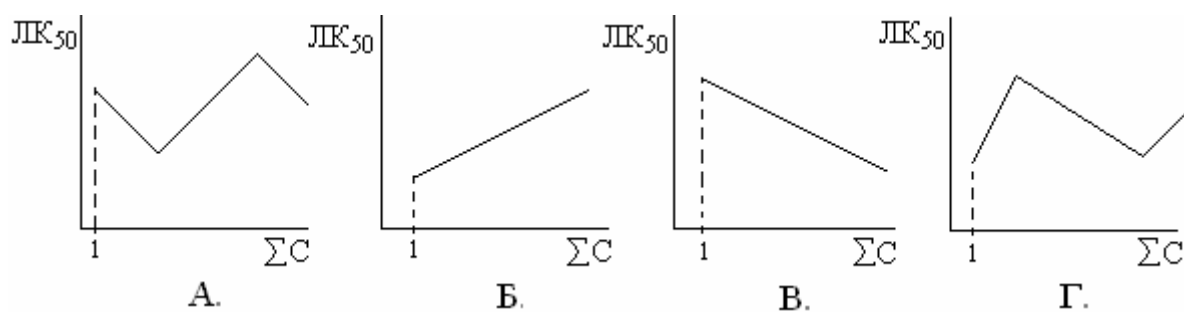
3.

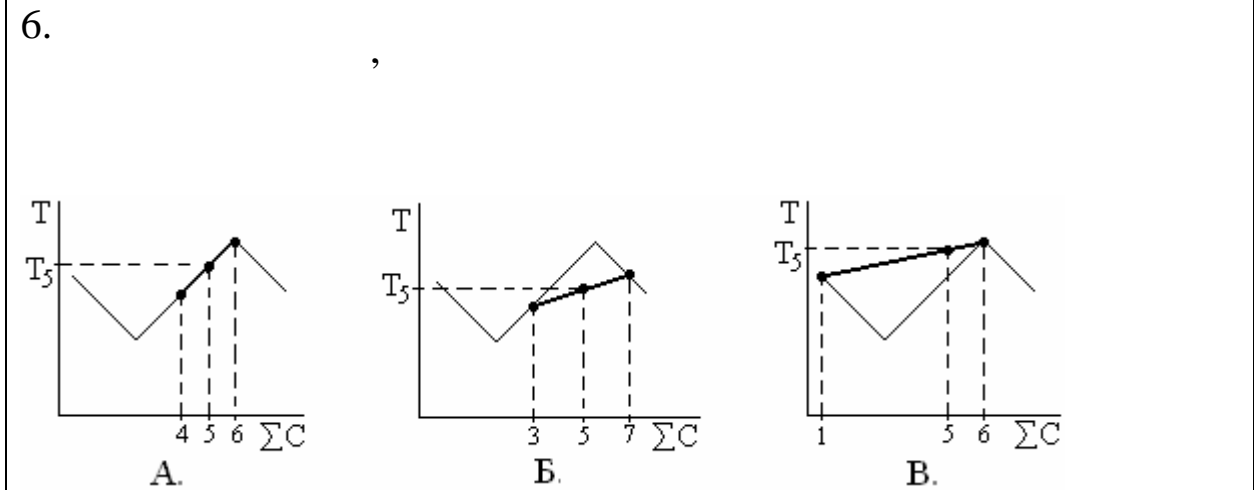
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4.



5.





7.

l_i

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8.

1.

2.

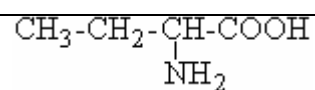
3.

4.

:

- 1, 3
- 1, 4
- 2, 4
- 2, 3

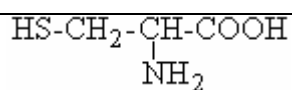
9.



,

- . -CH₃
- . -NH₂
- . -COOH
- . -CH₂-

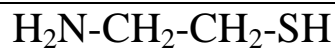
10.



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- . -SH
- . -CH₂-
- . -NH₂
- . -C=O
- . -COOH

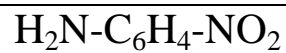
11.



,

- . -NH₂
- . -CH₂-
- . -CH₂-CH₂-
- . -SH

12.



,

- . -NH₂
- . >C₆H₄
- . -NO₂



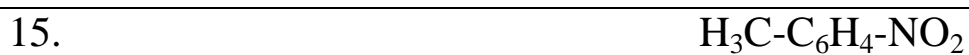
,

- . -S-
- . $-(\text{CH}_2)_2-$
- . $-\text{NH}_2$
- . -S-S-
- . -NH_2



,

- . $-\text{NH}_2$
- . $-\text{COOH}$
- . -S-
- . -S-S-
- . $\underset{|}{\text{CH}}-$
- . $-\text{CH}_2-$
- . -NH_2
- . $-\text{COOH}$



,

- . $>\text{C}_6\text{H}_4$
- . $-\text{CH}_3$
- . $-\text{NO}_2$

:

(, , ,)

1.

1.

2.
1,3)

3.

1,3)

$$\begin{aligned} & \cdot = \frac{\cdot 10 \cdot M}{\cdot m} \\ & (M/m < \cdot = \frac{1000 \cdot M}{\sum_i l_i} \\ & (M/m > \cdot = \frac{\cdot \cdot M}{\cdot M_x} \end{aligned}$$

2.

()

(m)

 M/m

1.

$$x = \frac{\cdot \cdot M}{\cdot M_x}$$

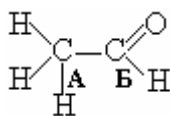
$$\cdot \frac{M}{m} > 1,3$$

2.

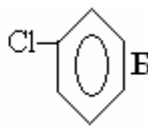
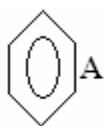
$$= \frac{\cdot 10 \cdot M}{\cdot m}$$

$$\cdot \frac{M}{m} < 1,3$$

3.

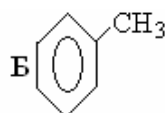
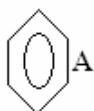
I. $l = 0,8$ /II. $l = 21273,9$ /

4.



I. $l = 1126,5$ / ; II. $l = 507,9$ / .

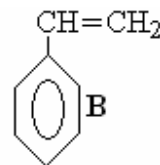
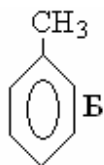
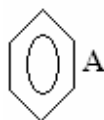
5.



I. $l = 1126,5$ /

II. $l = 507,9$ /

6.

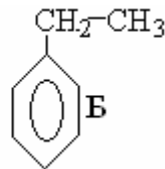
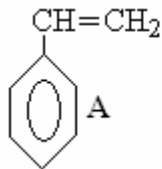


I. $l = 507,9$ /

II. $l = 1126,5$ /

III. $l = 7057,9$ /

7.

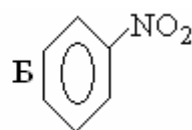
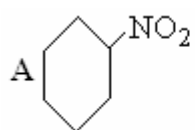


I. $l = 7057,9$ /

II. $l = 507,9$ /

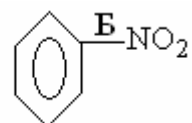
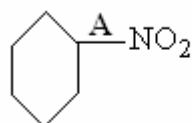
III. $l = 1126,5$ /

8.



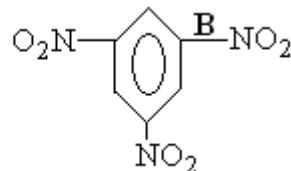
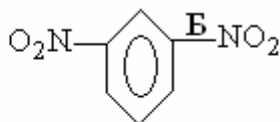
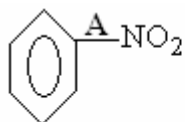
I. $l = 507,9$ / ; II. $l = 1126,5$ / ; III. $l = 173,7$ /

9.



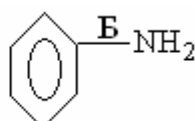
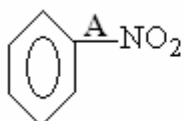
I. $l = 27970$ /
 II. $l = 119027,8$ /
 III. $l = 77851,5$ /

10.

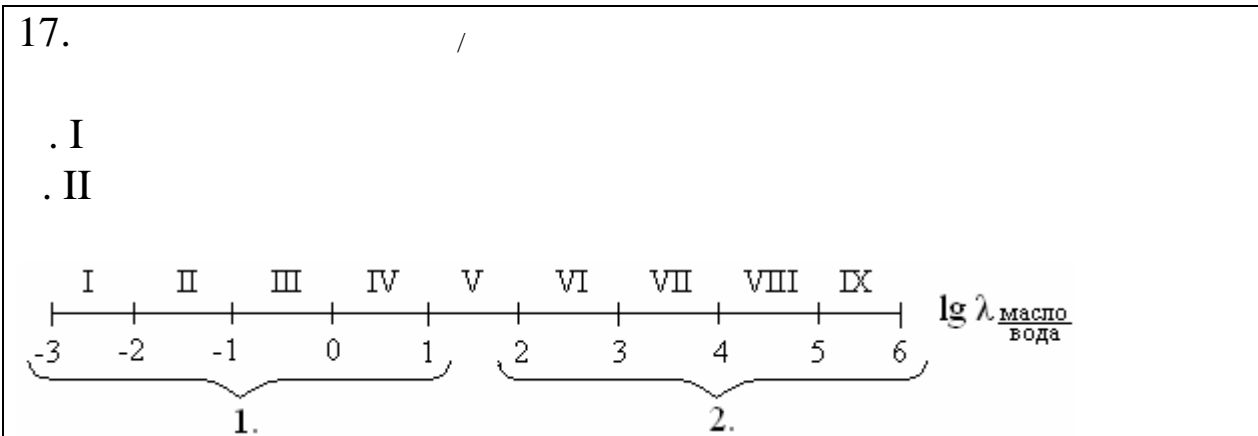
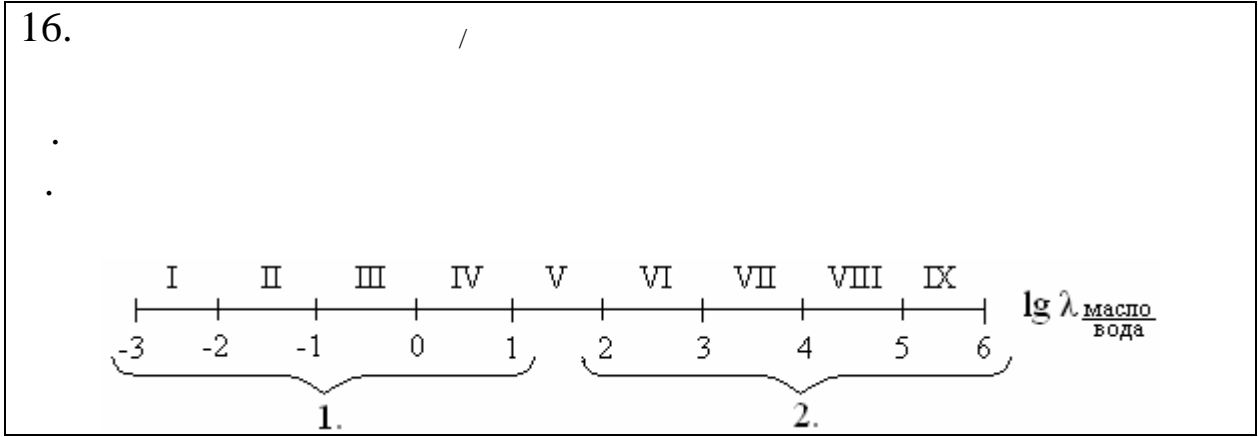


I. $l = 77851,5$ /
 II. $l = 66442$ /
 III. $l = 27970$ /
 IV. $l = 119027,8$ /

11.



I. $l = 27970$ /
 II. $l = 33302$ /
 III. $l = 6113,5$ /



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2. II	. 2000	II.

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1.	. $\lg p = 3,5 - 0,0202 \cdot (t + 3)$
2.	. $\lg D = 1 \pm (0 \div 1,1) + \frac{M}{115}$
3.	. $= \frac{1000 \cdot}{\sum_i l_i}$
4.	. $= \frac{10 \cdot \cdot \alpha}{\beta \cdot m}$

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 . $4d^{10}5s^25p^6$
 . $2p^63s^1$

4.

. 3^64s^1
 . $3s^23p^6$
 . $4d^{10}5s^25p^1$

5.

. $4d^{10}5s^25p^1$
 . $4d^{10}5s^25p^4$
 . $3d^{10}4s^24p^6$

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$$\text{I. } \lg = M - \dots \cdot t^\circ$$

$$\text{II. } \lg = M + \dots \cdot t^\circ$$

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$$\text{I. } \lg_{50} = M - \dots$$

$$\text{II. } \lg_{50} = M + \dots$$

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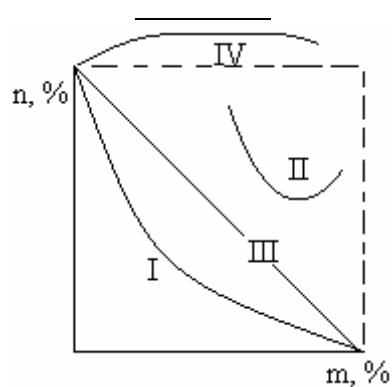
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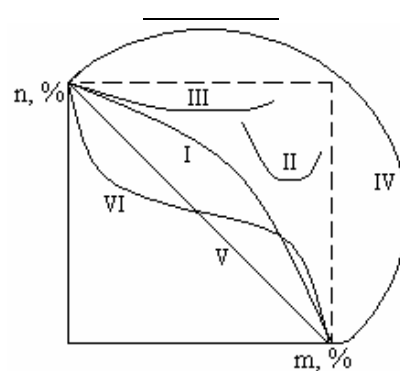
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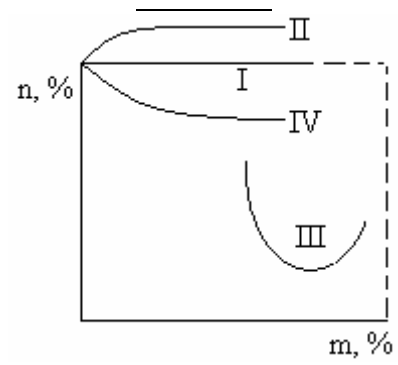
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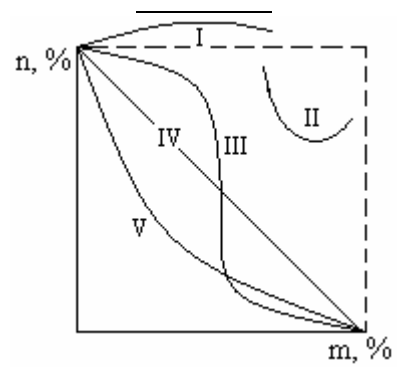
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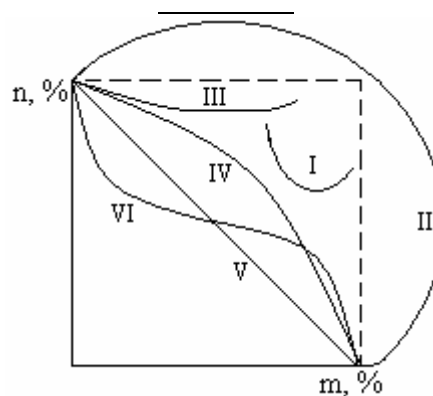
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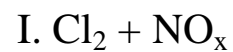
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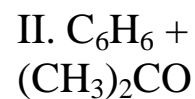


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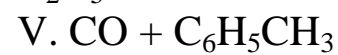
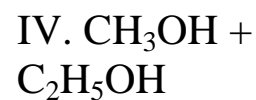
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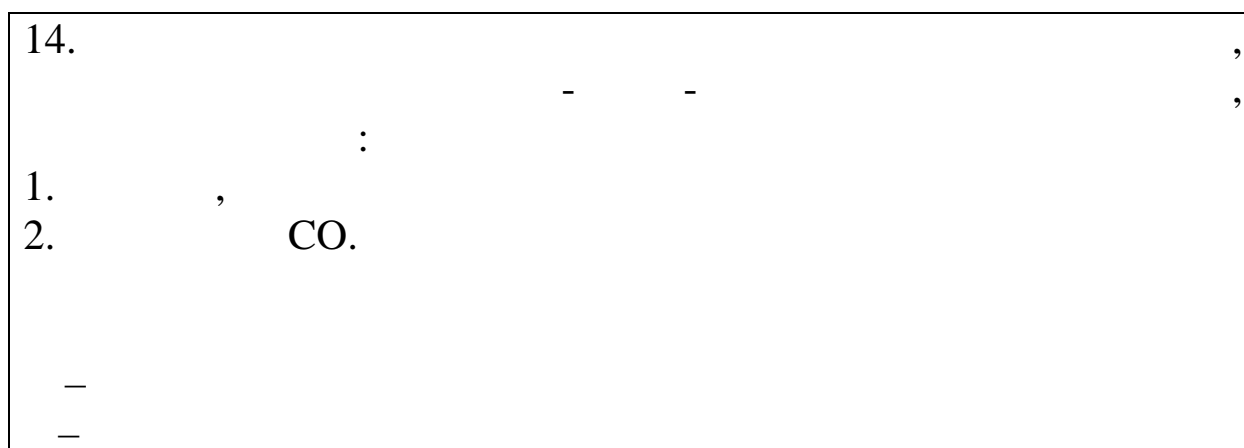
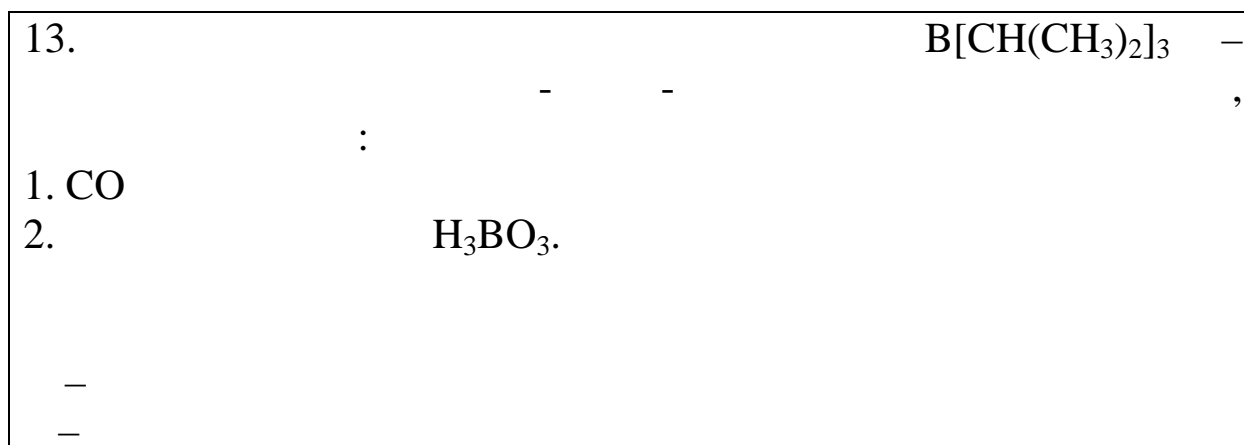
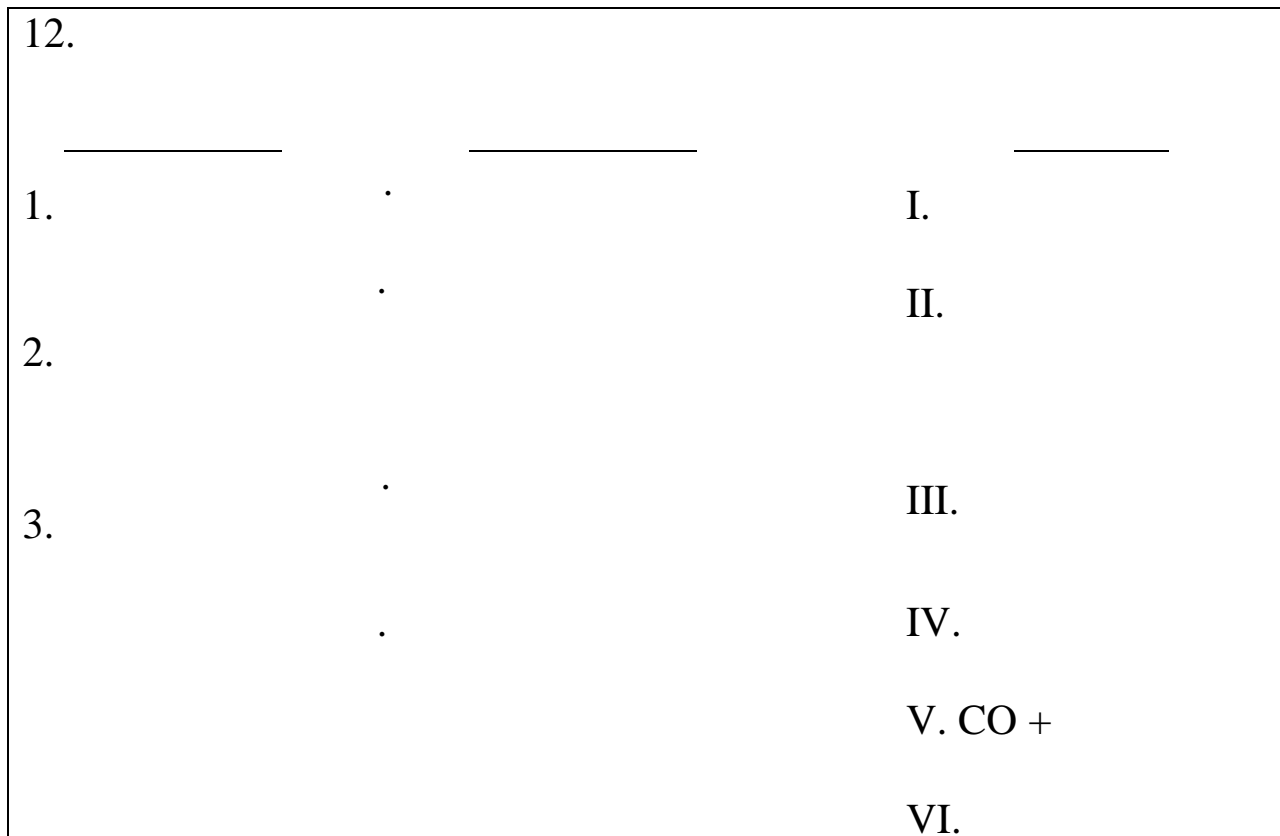


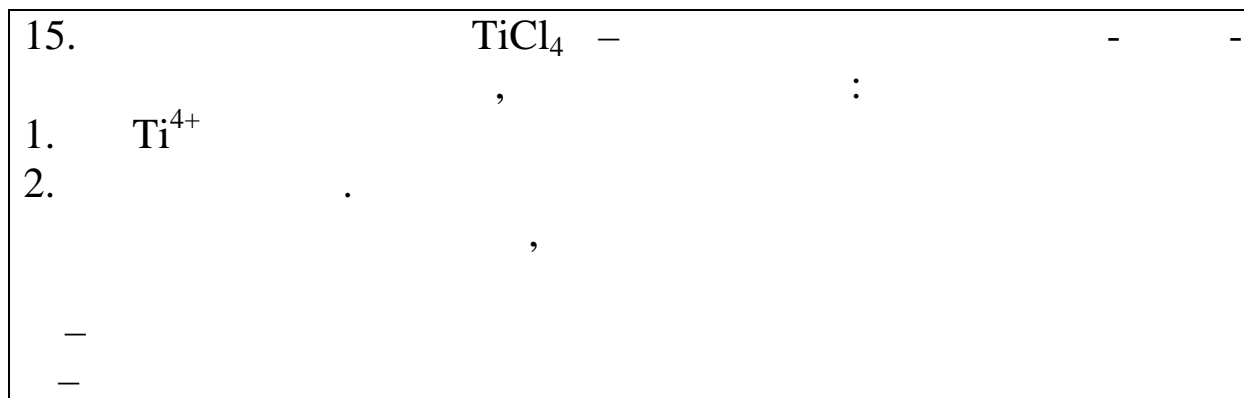
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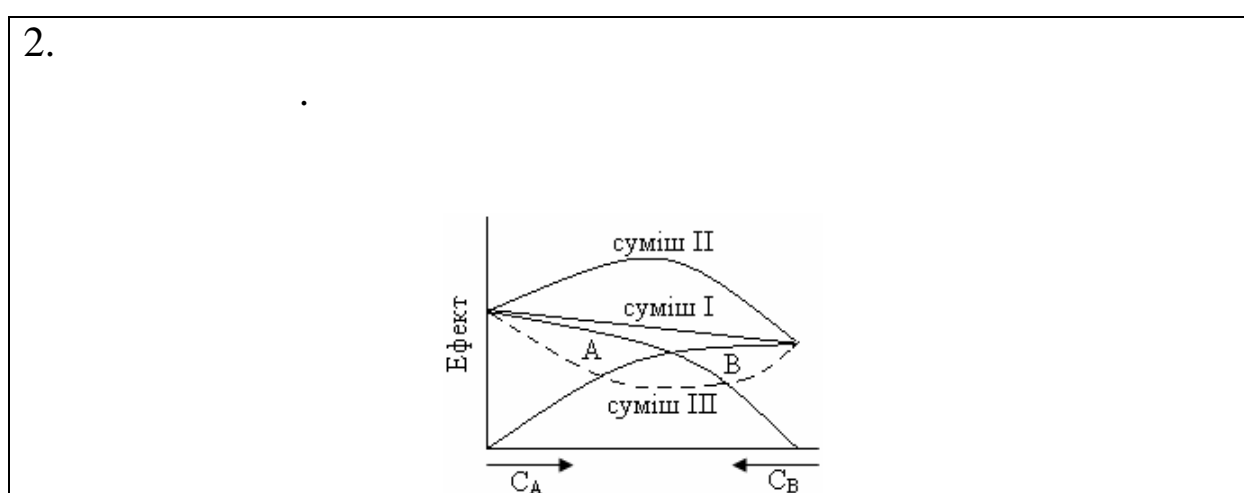
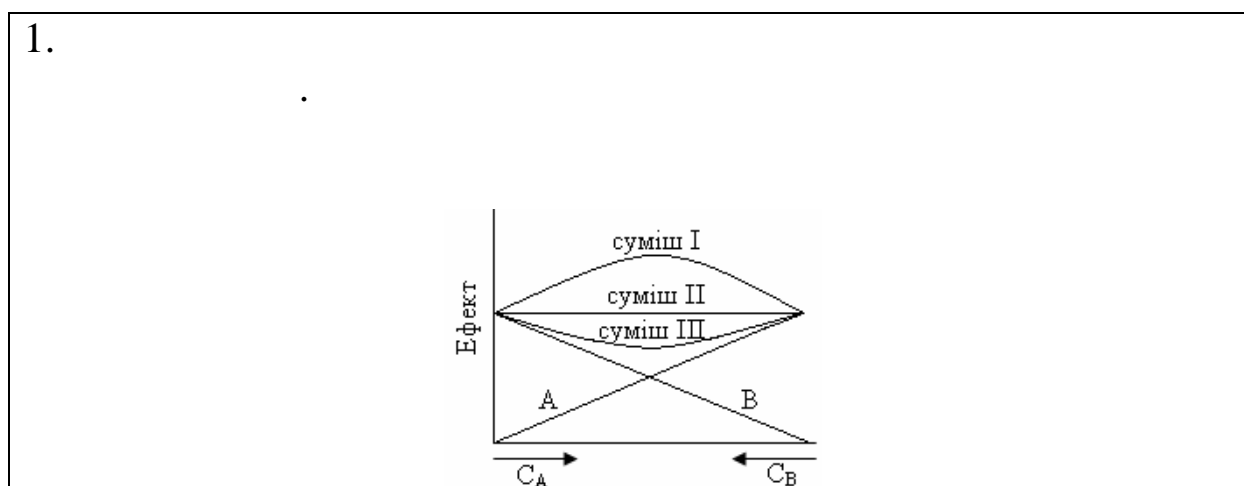
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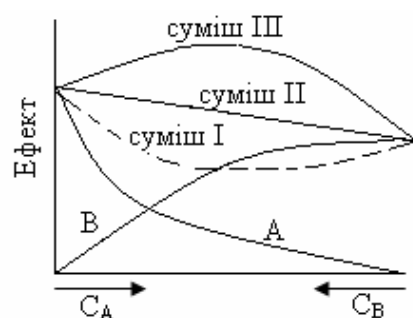




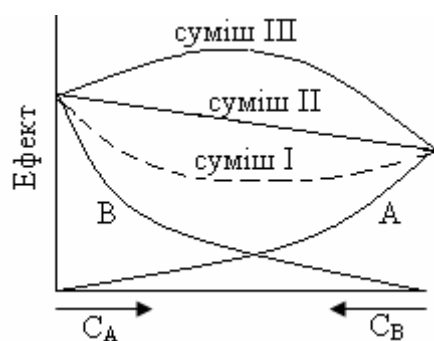
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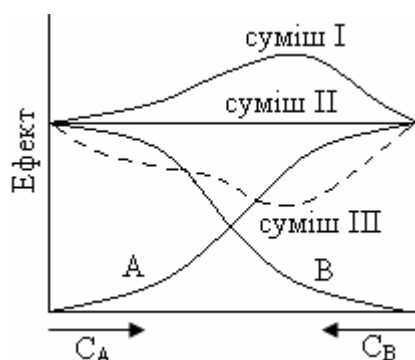
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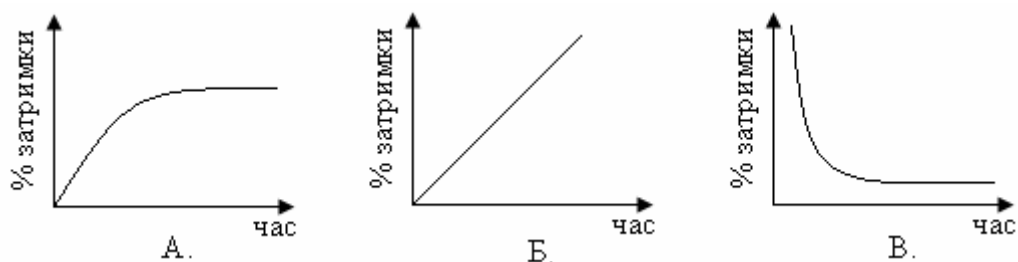
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4.	8/0,22	2	3	3
5.	7/0,2	-	-	7
6. 2	9/0,25	2	2	2+3*
7.	9/0,25	2	2	2+3*
8. ,	8/0,22	2	1	2+3*
9. ’	12/0,33	3	4	4+1*
10.	4/0,11	1	1	1+1*
11.	7/0,2	-	-	7
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. / : (057)700-38-64; 707-37-03, e-mail:
rio@khadi.kharkov.ua

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